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(54) Title: PYRIDINYL-2-PYRIMIDINAMINES USEFUL AS CARDIOTONICS AND PREPARATION THEREOF

#### (57) Abstract

(33) Priority Country:

2-(NB)-4-PY-5-Q-pyrimidines (I) and 2-(NB')-5-PY-4-Q'-pyrimidines (II) or salts thereof, where PY is 4- or 3-pyridinyl or 4- or 3-pyridinyl having one or two lower-alkyl substituents, Q and Q' are each hydrogen or methyl, NB is dimethylamino or N-(2-hydroxyethyl)methylamino and NB' is amino, dimethylamino, acetylamino or propionylamino; the preparation thereof; and their cardiotonic use.

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# PYRIDINYL-2-PYRIMIDINAMINES USEFUL AS CARDIOTONICS AND PREPARATION THEREOF

This invention relates to pyridinyl-2-pyrimidinamines useful as cardiotonics.

4-(4-Pyridinyl)-2-pyrimidinamine (no. 4a) and N-methyl-4-(4-pyridinyl)-2-pyrimidinamine (no. 4g) were reported by Bennett et al. [J. Med. Chem. 21, 623-628 (1978)] to be active when tested for antiinflammatory activity against carrageenan-induced edema in the rat but were said to be inactive when tested against adjuvant-induced edema in the rat. Other compounds reported active against carrageenan-induced edema were 4-(3-pyridinyl)-2-pyrimidinamine (4e), 5-methyl-4-(4-pyridinyl)-2-pyrimidinamine (4p), as well as the N-acetyl and N-propionyl derivatives (4h and 4i) of 4-(4-pyridinyl)-2-pyrimidinamine. Discussion of the antiinflammatory activity of the compounds disclosed in this publication concluded as follows:

"None of the compounds tested against adjuvantinduced edema in the rat displayed a level of
activity sufficient to warrant further investigation. Based on additional testing it would
appear that these compounds represent a series
of false positives in the carageenan-induced edema
model."

Compounds 4a and 4g of Bennett et al. were prepared by reacting 3-dimethylamino-1-(4-pyridinyl)-2-propen-1-one with guanidine or 1-methylguanidine, respectively. Compound 4e was prepared by reacting 3-dimethylamino-1-(3-pyridinyl)-2-propen-1-one with guanidine and compound 4p was prepared by reacting 2-dimethylamino-3-methyl-1-(4-pyridinyl)-2-propen-1-one with guanidine.



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Isomeric 2-(4- or 3-pyridinyl)-4-pyrimidinamines are shown as anti-allergic agents in U.S. Patents 4,086,233 and Reissue Patent 30,024, (reissue of U.S. Patent 4,032,523). N-Acetyl-2-(4-pyridinyl)-4-pyrimidinamine and other N-alkanoyl compounds are shown as intermediates in U.S. Patent 4,118,571.

The invention resides in a 4-PY-5-Q-2-pyrimidinamine or 5-PY-4-Q'-2-pyrimidinamine having the formula I or II respectively

or acid-addition salt thereof, where PY is 4- or 3-pyridinyl or 4- or 3-pyridinyl having one or two lower-alkyl substituents, Q and Q' are each hydrogen or methyl, NB is dimethylamino or N-(2-hydroxyethyl) methylamino and NB' is amino, dimethylamino, acetylamino or propionylamino. The compounds of formulas I and II are useful as cardiotonic agents, as determined by standard pharmacological evaluation procedures.

Preferred embodiments are those of formula I or II where PY is 4-pyridinyl or 3-pyridinyl, Q is hydrogen, Q' is methyl, NB is as defined above and NB' is amino.

The term "lower-alkyl" as used herein, e.g., as a substituent for PY, means alkyl radicals having from one to four carbon atoms which can be arranged as straight or branched chains, illustrated by methyl, ethyl, n-propyl, isopropyl, n-butyl, 2-butyl and isobutyl.



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The symbol "PY" as used herein, e.g., as the 4substituent in the pyrimidine ring of the compounds having
formula I or the 5-substituent in the pyrimidine ring
of the compounds having formula II, means 4- or 3-pyridinyl or 4- or 3-pyridinyl having one or two lower-alkyl
substituents, illustrated by 2-methyl-4-pyridinyl, 2,6dimethyl-4-pyridinyl, 3-methyl-4-pyridinyl, 2-methyl3-pyridinyl, 6-methyl-3-pyridinyl (alternatively named
2-methyl-5-pyridinyl), 2,3-dimethyl-4-pyridinyl, 2,6dimethyl-4-pyridinyl, 2-ethyl-4-pyridinyl, 2-isopropyl4-pyridinyl, 2-n-butyl-4-pyridinyl, 2,6-diethyl-4-pyridinyl, 2,6-diethyl-3-pyridinyl, 2,6-diisopropyl-4-pyridinyl, and the like.

The compounds of the invention having formula I or II are useful both in the free base form and in 15 the form of acid-addition salts, and both forms are within the purview of the invention. The acid-addition salts are simply a more convenient form for use; and in practice, use of the salt form inherently amounts to use of the base form. The acids which can be used to prepare 20 the acid-addition salts include preferably those which produce, when combined with the free base, pharmaceutically acceptable salts, that is, salts whose anions are relatively innocuous to the animal organism in pharmaceutical doses of the salts, so that the beneficial cardiotonic 25 properties inherent in the free base of the cardiotonically active compounds of the invention are not vitiated



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by side effects ascribable to the anions. In practicing the invention, it is convenient to use the free base form or the hydrochloric acid-addition salt; however, appropriate pharmaceutically acceptable salts within the scope of the invention are those derived from mineral acids such as sulfuric acid, phosphoric acid and sulfamic acid; and organic acids such as acetic acid, citric acid, lactic acid, tartaric acid, methanesulfonic acid, ethanesulfonic acid, benzenesulfonic acid, p-toluenesulfonic acid, cyclohexylsulfamic acid, quinic acid, and the like, which give the sulfate, phosphate, sulfamate, acetate, citrate, lactate, tartrate, methanesulfonate, ethanesulfonate, benzenesulfonate, p-toluenesulfonate, cyclohexylsulfamate and quinate, respectively.

The acid-addition salts of said basic compounds are prepared either by dissolving the free base in aqueous or aqueous-alcohol solution or other suitable solvents containing the appropriate acid and isolating the salt by evaporating the solution, or by reacting the free base and acid in an organic solvent, in which case the salt separates directly or can be obtained by concentration of the solution.

Although pharmaceutically acceptable salts of said basic compounds are preferred, all acid-addition salts are within the scope of our invention. All acid-addition salts are useful as sources of the free base



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form even if the particular salt per se is desired only as an intermediate product as for example when the salt is formed only for purposes of purification or identification, or when it is used as an intermediate in preparing a pharmaceutically acceptable salt by ion exchange procedures.

The molecular structures of the compounds of the invention were assigned on the basis of evidence provided by infrared, nuclear magnetic resonance and mass spectra, by the correspondence of calculated and found values for the elementary analyses, and, by their method of preparation.

The manner of making and using the instant invention will now be generally described so as to enable a person skilled in the art of pharmaceutical chemistry to make and use the same, as follows.

The compounds of formula I were prepared by heating 1,1-dimethylguanidine or 1-methyl-1-(2-hydroxyethyl)-guanidine, preferably as its acid-addition salt, e.g., sulfate or carbonate, with 3-dimethylamino-2-Q-1-PY-2-propen-1-one. When the guanidine salt is derived from a strong acid, e.g., sulfuric or hydrochloric acid, the reaction is run in the presence of, a base such as an alkali lower-alkoxide in a lower-alkanol, preferably sodium ethoxide or methoxide in refluxing ethanol or



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methanol. Although it can be used, no base is necessary using a guanidine salt of a weak acid, such as a carbonate or an acetate. Other solvents can be used, for example, n-propanol, 2-propanol, p-dioxane, tetrahydrofuran, dimethylformamide, 1,2-dimethoxyethane, and the like.

The intermediate 3-dimethylamino-2-Q-1-PY-2-propen-1-ones are generally known and are prepared by known means, for example, as described by Bennett et al. [J. Med. Chem. 21, 623-628 (1978)].

The compounds of formula II where NB' is amino or dimethylamino were prepared by heating guanidine or N,N-dimethylguanidine or salt thereof, e.g., sulfate or carbonate, with 3-dimethylamino-2-PY-3-Q'-2-propen-l-al. This reaction can be run as described above for the preparation of the compounds of formula I by reacting said guanidine or salt with 3-dimethylamino-3-Q-1-PY-2-propen-l-one.

The intermediate 3-dimethylamino-2-PY-3-Q'-2-propen-1-als are generally known and are prepared by known means (U.S. Patent 4,004,012, issued January 18, 1977).

The compounds of formula II where NB' is acetylamino or propionylamino are prepared by reacting the compound of formula II where NB' is amino with an acetylating or propionylating agent, preferably acetic or propionic anhydride, conveniently by heating the reactants in a suitable aprotic solvent, e.g., pyridine, p-dioxane, tetrahydrofuran, dimethylformamide, 1,2-dimethoxyethane, and the like.

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The following examples will further illustrate the invention without, however, limiting it thereto.

#### A. 4-PY-5-Q-2-PYRIMIDINAMINES

- A-1. N,N-Dimethyl-4-(3-pyridinyl)-2-pyrimidinamine - To a solution containing 2.3 g of sodium dissolved 5 in 200 ml of ethanol was added 13.6 g of 1,1-dimethylguanidine sulfate and 17.6 g of 3-dimethylamino-1-(3pyridinyl)-2-propen-1-one and the resulting mixture was refluxed for 3 hours and allowed to cool. The solid was filtered off and the filtrate was evaporated to dry-10 ness. The residue was taken up in isopropyl alcohol, the solution made acidic with acetic acid and filtered. The filtrate was evaporated to dryness and taken up in ether. The ether layer was treated with decolorizing charcoal, washed several times with 5% aqueous potassium 15 bicarbonate solution and then treated with ethanolic hydrogen chloride, stirred well and the separated solid The solid was recrystallized from a minimum quantity of hot methanol followed by addition of ethanol and cooling. The precipitated product was collected and dried to produce 9.2 g of N,N-dimethyl-4-(3-pyridinyl)-2-pyrimidinamine as its dihydrochloride, m.p. 230-241°C.
  - A-2. N-(2-Hydroxyethyl)-N-methyl-4-(4-pyridinyl)-2-pyrimidinamine - To a solution containing 5 g of sodium dissolved in 300 ml of ethanol was added 20 g of N-(2hydroxyethyl)-N-methylguanidine sulfate and 17.6 g of



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3-dimethylamino-1-(4-pyridinyl)-2-propen-1-one and the mixture was heated with stirring on a steam bath for The reaction mixture was made acidic with acetic acid, next made basic with ammonium hydroxide and the resulting solution was evaporated to dryness. The residue was extracted with ether and then with warm isopropyl acetate. The combined extracts were chilled and the separated solid was collected. The solid was dissolved in methylene dichloride, treated with decolorizing charcoal and filtered. The filtrate was evaporated to dryness, the residue dissolved in isopropyl alcohol, and the alcohol solution treated with excess hydrogen chloride in absolute ethanol. The separated solid was collected, washed successively with isopropyl alcohol, hot isopropyl acetate and ether, and dried to produce 22.5 g of N-(2-hydroxyethyl)-N-methyl-4-(4-pyridinyl)-2-pyrimidinamine as its dihydrochloride, m.p. 191-195°C.

guanidine sulfate was prepared as follows: A mixture

containing 83.6 g of methyl isothiourea sulfate and 90

g of N-(2-hydroxyethyl)methylamine was heated on a steam

bath under vacuum until no more methyl mercaptan was

evolved. The solid remaining was boiled successively

with hot isopropyl alcohol, hot absolute ethanol and

then ether, and collected to yield 91 g of crystalline

N-(2-hydroxyethyl)-N-methylguanidine sulfate.



- 2-pyrimidinamine as its dihydrochloride hemihydrate,
  m.p. 201-205°C., was prepared following the procedure
  described in Example A-2 using 5.5 g of sodium in 300
  ml of absolute ethanol, 22 g of N-(2-hydroxyethyl-N-methyl guanidine sulfate, 17.6 g of 3-dimethylamino-1-(3-pyridinyl)-2-propen-1-one and a reflux period of 90 minutes.
- A-4. N,N-Dimethyl-4-(4-pyridinyl)-2-pyrimidinamine - To 500 ml of absolute ethanol was added 26.4 g of 3-dimethylamino-1-(4-pyridinyl)-2-propen-1-one, 10 19.1 g of 1,1-dimethylguanidine sulfate and 7.6 g of sodium methoxide. The reaction mixture was refluxed for 3 hours, allowed to cool to room temperature and The filtrate was evaporated to remove most filtered. of the ethanol and was then treated with several volumes 15 of ether. The resulting precipitate was filtered off and the filtrate concentrated in vacuo to yield 23.6 g of viscous orange oil. The oil was transferred to a silica gel bed (5 inches in diameter and 3 inches high) 20 The bed was eluted with with the aid of 20 ml ether. ether and the ether fractions (all containing only one spot) were combined and evaporated to dryness. The remaining viscous oil, on chilling and scratching with a glass rod, crystallized. The solid was broken up, triturated 25 with a small volume of anhydrous ether, collected and dried in a vacuum dessicator to remove last traces of



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solvent from the silica gel treatment to yield 16.3 g of hygroscopic N,N-dimethyl-4-(4-pyridinyl)-2-pyrimidin-amine, m.p. 44-47[C.

Acid-addition salts of N,N-dimethyl-4-(4-pyridinyl)-2-pyrimidinamine are conveniently prepared by adding 5 to a mixture of 1 g of N,N-dimethyl-4-(4-pyridinyl)-2pyrimidinamine in about 20 ml of aqueous methanol the appropriate acid, e.g., methanesulfonic acid, concentrated sulfuric acid, concentrated phosphoric acid, to a pH of about 2 to 3, chilling the mixture after partial evapor-10 ation and collecting the precipitated salt, e.g., dimethanesulfonate, sulfate, phosphate, respectively. Also, the acid-addition salt is conveniently prepared in aqueous solution by adding to water with stirring molar equivalent quantities each of N,N-dimethyl-4-(4-pyridinyl)-15 2-pyrimidinamine and the appropriate acid, e.g., lactic acid or hydrochloric acid, to prepare respectively the lactate or hydrochloride salt of N,N-dimethyl-4-(4-pyridinyl)-2-pyrimidinamine in aqueous solution.

A-5. N,N,5-Trimethyl-4-(4-pyridinyl)-2-pyrimidinamine, m.p. 114-116°C., 9.5 g, was obtained following
the procedure described in Example A-5 using 13.6 g of
N,N-dimethyl guanidine sulfate, 20.9 g of 3-dimethylamino2-methyl-1-(4-pyridinyl)-2-propen-1-one, 5.4 g of sodium
methoxide, 380 ml of absolute ethanol and a reflux period
of 17 hours.



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Acid-addition salts of N,N,5-trimethyl-4-(4-pyridinyl)-2-pyrimidinamine are conveniently prepared by adding to a mixture of 1 g of N,N,5-trimethyl-4-(4-pyridinyl)-2-pyrimidinamine in about 20 ml of aqueous methanol the appropriate acid, e.g., methanesulfonic acid, concentrated sulfuric acid, concentrated phosphoric acid, to a pH of about 2 to 3, chilling the mixture after partial evaporation and collecting the precipitated salt, e.g., dimethanesulfonate, sulfate, phosphate, respectively.

Also, the acid-addition salt is conveniently prepared in aqueous solution by adding to water with stirring molar equivalent quantities each of N,N,5-trimethyl-4-(4-pyridinyl)-2-pyrimidineamine and the appropriate acid, e.g., lactic acid or hydrochloric acid, to prepare respectively the lactate or hydrochloride salt of N,N,5-trimethyl-4-(4-pyridinyl)-2-pyrimidinamine in aqueous solution.

Following the procedure of Example A-1 but using in place of 1,1-dimethylguanidine sulfate and 3-dimethyl-amino-1-(3-pyridinyl)-2-propen-1-one molar equivalent quantities of the appropriate guanidine derivative and 3-dimethylamino-2-Q-1-PY-2-propen-1-one, it is contemplated that the corresponding 4-PY-5-Q-2-pyrimidinamines of Examples A-6 through A-9 can be obtained.

A-6. N-(2-Hydroxyethyl)-N-methyl-4-(2-methyl-3-pyridinyl)-2-pyrimidinamine, using N-(2-hydroxyethyl)-N-methylguanidine sulfate and 3-dimethylamino-1-(2-methyl-3-pyridinyl)-2-propen-1-one.



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- A-7. N,N,5-Trimethyl-4-(5-methyl-3-pyridinyl)2-primidinamine, using N,N-dimethylguanidine sulfate
  and 3-dimethylamino-2-methyl-1-(5-methyl-3-pyridinyl)2-propen-1-one.
- A-8. 4-(3-Ethyl-4-pyridinyl)-N,N-dimethyl-2-pyrimidinamine, using 1,1-dimethylguanidine sulfate and 1(3-ethyl-4-pyridinyl)-3-dimethylamino-2-propen-1-one.
- A-9. N,N-Dimethyl-4-(2,6-dimethyl-4-pyridinyl)
  2-pyrimidinamine, using 1,1-dimethylguanidine sulfate

  and 3-dimethylamino-1-(2,6-dimethyl-4-pyridinyl)-2-propen
  1-one.

## B. 5-PY-4-Q'-2-PYRIMIDINAMINES

B-1. 5-(4-Pyridinyl)-2-pyrimidinamine - A mixture containing 20 g of 3-dimethylamino-2-(4-pyridinyl)-2
propen-1-al, 25 g of guanidine carbonate and 100 ml of ethanol was refluxed for 4 hours and concentrated in vacuo to remove the solvent. The residue was stirred in ice cold water and the crystalline product was collected, washed with water and dried to produce 12.1 g of 5-(4-pyridinyl)-2-pyrimidinamine, m.p. 210-212°C.

Acid-addition salts of 5-(4-pyridinyl)-2-pyrimidinamine are conveniently prepared by adding to a mixture
of 1 g of 5-(4-pyridinyl)-2-pyrimidinamine in about 20
ml of aqueous methanol the appropriate acid, e.g., methanesulfonic acid, concentrated sulfuric acid, concentrated
phosphoric acid, to a pH of about 2 to 3, chilling the
mixture after partial evaporation and collecting the



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precipitated salt, e.g., dimethanesulfonate, sulfate, phosphate, respectively. Also, the acid-addition salt is conveniently prepared in aqueous solution by adding to water with stirring molar equivalent quantities each of 5-(4-pyridinyl)-2-pyrimidinamine and the appropriate acid, e.g., lactic acid or hydrochloric acid, to prepare respectively the lactate or hydrochloride salt of 5-(4-pyridinyl)-2-pyrimidinamine in aqueous solution.

- B-2. N.N-Dimethyl-5-(4-pyridinyl)-2-pyrimidin
  amine, m.p. 176-178°C., 5.8 g, was prepared following
  the procedure described in Example B-1 using 17.6 g of
  3-dimethylamino-2-(4-pyridinyl)-2-propen-1-al, 27.2 g
  of 1,1-dimethylguanidine sulfate, 10.8 g of sodium methoxide, 100 ml of ethanol and recrystallization of the
  product from ethanol.
- B-3. N,N-Dimethyl-5-(3-pyridinyl)-2-pyrimidinamine A mixture containing 17 g of 3-dimethylamino2-(3-pyridinyl)-2-propen-1-al, 13.6 g of 1,1-dimethylguanidine sulfate, 16 g of sodium methoxide and 500 ml of
  methanol was refluxed for 3 hours and then allowed to
  stand at room temperature overnight. The reaction mixture was evaporated in vacuo to remove the solvent and
  water was added to the residue. The solid was collected,
  washed with water and dried. The solid was dissolved
  in 6N hydrochloric acid, the solution filtered through
  diatomaceous earth and the filtrate treated with isopropyl



alcohol and cooled. The separated product was collected, washed successively with isopropyl alcohol and ether, and dried to produce 4 g of N,N-dimethyl-5-(3-pyridinyl)-2-pyrimidamine as its hydrochloride, m.p. 295°C. with decomposition.

B-4. 4-Methyl-5-(4-pyridinyl)-2-pyrimidinamine A mixture containing 15.6 g of 3-dimethylamino-3-methyl2-(4-pyridinyl)-2-propen-1-al, 17.7 g of guanidine carbonate, 5.4 g of sodium methoxide and 150 ml of absolute
ethanol was stirred at room temperature for 6 hours and
then heated on a steam bath for about 20 minutes and
cooled. The reaction mixture was stripped in vacuo and
the resulting solid residue was slurried with water,
the mixture acidified using acetic acid and put in a
refrigerator overnight. The separated solid was collected,
dried in a vacuum oven at 60°C., recrystallized from
hot ethanol and dried in vacuo at 95°C. to produce 9.6
g of 4-methyl-5-(4-pyridinyl)-2-pyrimidinamine, m.p.
215-217°C.

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B-5. N-[5-(4-Pyridinyl)-2-pyrimidinyl]acetamide A mixture containing 17.2 g of 5-(4-pyridinyl)-2-pyrimidamine, 61.2 g of acetic anhydride and 200 ml of pyridine
was refluxed for 2 hours and cooled. The separated solid
was collected, washed successively with ethanol and ether,
and then dried. The solid was recrystallized from dimethylformamide, washed successively with ethanol and ether,
and dried to produce 10 g of N-[5-(4-pyridinyl)-2-pyrimidinyl]acetamide, m.p. 261-264°C.



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N-[5-(4-Pyridinyl)-2-pyrimidinyl]propanamide -A mixture containing 17 g of 5-(4-pyridiny1)-2-pyrimidinamine, 78 g of propionic anhydride and 500 ml of chloroform was refluxed for 90 minutes after which time no apparent reaction had taken place. To the reaction mixture was added 30 ml of pyridine and refluxing was continued for 4 hours, after which time no apparent reaction had taken place. The chloroform was then stripped off in vacuo, 200 ml of pyridine was added and the reaction mixture was refluxed for 40 hours and cooled. The separated solid was collected, washed with ethyl acetate and dried. The filtrate was stripped in vacuo and the resulting residue was treated with cold water, collected, washed with a small quantity of water and dried. The solids were combined, recrystallized from dimethylformamide, 15 washed successively with ethanol and ether and dried to produce 9.5 g of N-[5-(4-pyridinyl)-2-pyrimidinyl]propanamide, m.p. 220-223°C.

Acid-addition salts of N-[5-(4-pyridinyl)-2-pyrimidinyl]propanamide are conveniently prepared by adding to a mixture of 1 g of N-[5-(4-pyridiny1)-2-pyrimidiny1]propanamide in about 20 ml of aqueous methanol the appropriate acid, e.g., methanesulfonic acid, concentrated sulfuric acid, concentrated phosphoric acid, to a pH of about 2 to 3, chilling the mixture after partial evaporation and collecting the precipitated salt, e.g., dimethane-



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sulfonate, sulfate, phosphate, respectively. Also, the acid-addition salt is conveniently prepared in aqueous solution by adding to water with stirring molar equivalent quantities each of N-[5-(4-pyridinyl)-2-pyrimidinyl]-propanamide and the appropriate acid, e.g., lactic acid or hydrochloric acid, to prepare respectively the lactate or hydrochloride salt of N-[5-(4-pyridinyl)-2-pyrimidinyl]-propanamide in aqueous solution.

using in place of 3-dimethylamino-2-(4-pyridinyl)-2-propenl-al or 3-dimethylamino-2-(4-pyridinyl)-2-propen-l-al
and 1,1-dimethylguanidine sulfate or guanidine carbonate
molar equivalent quantities of the appropriate 3-dimethylamino-3-Q'-2-py-2-propen-l-al and guanidine derivative,
it is contemplated that the corresponding compounds of
Examples B-7 through B-1l can be obtained.

- B-7. <u>5-(3-Pyridinyl)-2-pyrimidinamine</u>, using 3-dimethylamino-2-(3-pyridinyl)-2-propen-1-al and guanidine carbonate.
- B-8. 5-(2-Methyl-3-pyridinyl)-2-pyrimidinamine, using 3-dimethylamino-2-(2-methyl-3-pyridinyl)-2-propen-1-al and guanidine carbonate.
  - B-9. 5-(2-Methyl-4-pyridinyl)-2-pyrimidinamine, using 3-dimethylamino-2-(2-methyl-4-pyridinyl)-2-propen-1-al and guanidine sulfate.
    - B-10. 4-Methyl-5-(5-methyl-3-pyridinyl)-2-pyrimidinamine, using 3-dimethylamino-3-methyl-2-(5-methyl3-pyridinyl)-2-propen-1-al and guanidine sulfate.

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N, N-Dimethyl-5-(2,6-dimethyl-4-pyridinyl)-2-pyrimidinamine, using 3-dimethylamino-2-(2,6-dimethyl-4-pyridinyl)-2-propen-1-al and 1,1-dimethylguanidine sulfate.

The usefulness of the compounds of formulas I and II, or pharmaceutically acceptable acid-addition salts thereof, as cardiotonic agents is demonstrated by their effectiveness in standard pharmacological test procedures, for example, in causing a significant increase in the contractile force of the isolated cat or guinea 10 pig atria and papillary muscle and/or in causing a significant increase in the cardiac contractile force in the anesthetized dog with low or minimal changes in heart rate and blood pressure. Detailed descriptions of these test procedures appear in U.S. Patent 4,072,746, issued February 7, 1978.

When tested by the isolated cat or guinea pig atria and papillary muscle procedure, the compounds of formulas I and II, or pharmaceutically acceptable acidaddition salts thereof at doses of 10, 30, 100 and/or 300  $\mu$ g/ml, were found to cause significant increases, that is, greater than 25% (cat) or 30% (guinea pig) in papillary muscle force and significant increases, that is, greater than 25% (cat) or 30% (guinea pig), in right atrial force, while causing a lower percentage increase in right atrial rate. Because of the lower control active



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tensions of guinea pig tissues, the percent change from control values of both rate and force responses is elevated slightly, i.e., 5%. Thus, whereas cardiotonic activity is ascertained with a papillary muscle force or right atrial force increase of 26% and greater in the cat test, corresponding activity in the guinea pig test is designated with a papillary muscle force or right atrial force increase of 31% or greater. For example, when tested at one or more said dose levels by this procedure in the cat or guinea pig test, the compounds of the invention were found to cause respective increases in papillary muscle force (PMF) and right atrial force (RAF) given in Table A.

Table A

			Idate w					
15			In Vitro Cardiotonic Activity					
		Con	Dose ug/ml	Percentage Increase				
	Example	Cat or g.p.		RARa	RAF	PMFC	Nq	
	A-1	cat	100	48	57	104	3/4	
20	A-2	cat	30.	20	26	38	3/5	
			100	44	51	66	3/5	
	A-3	g.p.	30	10	32	56	6/8	
	A 3	3.5.	100	19	118	127	6/8	
	A-4	cat	30	30	34	63	3/5	
	Ач	04.0	100	30	107	179	3/5	
	A-5	g.p.	100	0	36	68	3/4	
	B-1	cat	30	8	26	26	2/3	
	<b>9 4</b>	.,000	100	12	40	79	2/3	
	B-2	g.p.	100	41	279	51	3/5	
	B-3	cat	100	33	57	63	2/2	
	B-4	g.p.	10	20	50	84	3/5	
	<b>D</b> - 4	3.5.	30	49	96	147	3/5	
		•	100	60	191	198	3/5	
25	B-5	cat	100	15	31	54	2/5	
			300	15	58	36	2/2	
	B-6	cat	300				-, - 	

a) Right atrial rate.

b) Right atrial force.

c) Papillary muscle force.

d) Number of preparations.

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When tested by said anesthetized dog procedure, the compounds of formula I or II at doses of 3.0 and/or 10.0 mg/kg administered intravenously were found to cause significant increases, that is, 25% or greater, in cardiac contractile force or cardiac contractility with lower changes in heart rate and blood pressure. example, when tested at one or more of said dose levels by this procedure, the compounds of Examples B-2, B-5 and B-6 were found to cause increases of about 30% to 90% in contractile force and lower changes in heart rate. and blood pressure.

A cardiotonic composition comprises a pharmaceutically acceptable carrier and, as the active component thereof, the cardiotonic compound of formula I or II or pharmaceu-15 tically acceptable acid-addition salt thereof. A method for increasing cardiac contractility in a patient requiring such treatment comprises administering to such patient a cardiotonically effective amount of the compound of formula I or II or pharmaceutically acceptable acid-addition salt thereof. In clinical practice said compound or salt thereof will normally be administered orally or parenterally in a wide variety of dosage forms.

Solid compositions for oral administration include compressed tablets, pills, powders and granules. such solid compositions, at least one of the active com-25 pounds is admixed with at least one inert diluent such



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as starch, calcium carbonate, sucrose or lactose. These compositions may also contain additional substances other than inert diluents, e.g., lubricating agents, such as magnesium stearate, talc, and the like.

Liquid compositions for oral administration include pharmaceutically acceptable emulsions, solutions, suspensions, syrups and elixirs containing inert diluents commonly used in the art, such as water and liquid paraffin. Besides inert diluents such compositions may also contain adjuvants, such as wetting and suspending agents, and sweetening, flavoring, perfuming and preserving agents. According to the invention, the compounds for oral administration also include capsules of absorbable material, such as gelatin, containing said active component with or without the addition of diluents or excipients.

Preparations according to the invention for parenteral administration include sterile aqueous, aqueousorganic, and organic solutions, suspensions and emulsions.
Examples of organic solvents or suspending media are
propylene glycol, polyethylene glycol, vegetable oils
such as olive oil and injectable organic ester such as
ethyl oleate. These compositions can also contain adjuvants such as stabilizing, preserving, wetting, emulsifying and dispersing agents.



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They can be sterilized, for example by filtration through a bacterial-retaining filter, by incorporation of sterilizing agents in the compositions, by irradiation or by heating. They can also be manufactured in the form of sterile solid compositions which can be dissolved in sterile water or some other sterile injectable medium immediately before use.

The percentage of active component in the said composition and method for increasing cardiac contractility can be varied to that a suitable dosage is obtained. The dosage administered to a particular patient is variable, depending upon the clinician's judgement using as the criteria: the route of administration, the duration of treatment, the size and condition of the patient, the potency of the active component and the patient's response thereto. An effective dosage amount of active component can thus only be determined by the clinician considering all criteria and utilizing his best judgement on the patient's behalf.



# CLAIMS.

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#### 1. A compound having the formula

or an acid-addition salt thereof, where PY is 4- or 3pyridinyl or 4- or 3-pyridinyl having one or two lower-alkyl
substituents, Q and Q' are each hydrogen or methyl, NB is
dimethylamino or N-(2-hydroxyethyl)methylamino and NB' is
amino, dimethylamino, acetylamino or propionylamino.

- 2. A compound according to claim 1, where PY is 4-pyridinyl or 3-pyridinyl and Q is hydrogen.
- 3. A compound according to claim 1 of Formula I where PY is 4-pyridinyl, Q is hydrogen and NB is dimethylamino.
- 4. A compound according to claim 1 of Formula I, where PY is 3-pyridinyl, Q is hydrogen and NB is dimethylamino.
- 5. A compound according to claim 1 of Formula I where PY is 3-pyridinyl, Q is hydrogen and NB is N-(2-hydroxyethyl)methylamino.
- 6. A compound according to claim 1 of Formula II where PY is 4-pyridinyl, Q' is methyl and NB' is amino.
- 7. A process for preparing a compound according to claim 1, which comprises
  - a. heating 1,1-dimethyl guanidine or 1-methyl-1(2-hydroxyethyl)-guanidine, or an acid-addition
    salt of said guanidine, with 3-dimethylamino-2-Q1-PY-2-propen-1-one to prepare a compound of
    Formula I, or
  - b. heating quanidine or N,N-dimethyl-quanidine, or an acid-addition salt of said quanidine, with 3-dimethyl-amino-2-PY-3-Q'-2-propen-1-al to prepare a compound of Formula II where NB' is amino or dimethylamino, and, if desired, reacting a compound



of Formula II obtained where NB' is amino with an acetylating or propionylating agent to prepare a compound where NB' is acetylamino or propionylamino, and, if desired, converting a compound obtained in its free base form to an acid-addition salt thereof.

- 8. A cardiotonic composition for increasing cardiac contractility, said composition comprising a pharmaceutically acceptable inert carrier and, as the active component thereof, a cardiotonically effective amount of a compound according to any one of claims 1-6 or a pharmaceutically acceptable acid-addition salt thereof.
- 9. A compound for increasing cardiac contractility in a patient requiring such treatment by oral or parenteral administration which comprises a compound according to any one of claims 1-6 or a pharmaceutically acceptable acidaddition salt thereof.



#### INTERNATIONAL SEARCH REPORT

International Application No PCT/US84/01051

		M OF SUBJECT MATTER (if several class					
According to International Patent Classification (IPC) or to both National Classification and IPC Int. Cl. CO7 D 401/04; A61K 31/505							
II. FIELD	S SEARCH	1ED					
		Minimum Docume	entation Searched 4				
Classification	on System		Classification Symbols				
U. S. 544/331; 424/251		544/331; 424/251					
		Documentation Searched other to the Extent that such Document	than Minimum Documentation s are included in the Fields Searched 5				
Che	Chemical Abstracts, Vol. 1-100, pyrimidine, 1,3-diazine						
III. DOCU	MENTS C	ONSIDERED TO BE RELEVANT 14					
Category •	Citati	on of Document, 16 with indication, where ap	propriate, of the relevant passages 17	Relevant to Claim No. 18			
A	US,	4,086,233, issued 2 Lesher e		1-6,8-9			
. I	US,	4,118,571, issued 3 Lesher e	October 1978, t al.	1-6,8-9			
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х	J. Med. Chem. 21(7), 1978, Bennet et al., pages 623-628. Synthesis and Antiinflammatory Activity of Trisubstituted Pyrimidines and Trialzines						
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<ul> <li>Special categories of cited documents: 15</li> <li>"A" document defining the general state of the art which is not considered to be of particular relevance</li> <li>"E" earlier document but published on or after the international filing date</li> <li>"L" document which may throw doubts on priority claim(s) or which is cited to establish the publication date of another citation or other special reason (as specified)</li> <li>"O" document referring to an oral disclosure, use, exhibition or other means</li> <li>"P" document published prior to the international filing date but later than the priority date claimed</li> <li>"T" later document published after the international filing date or priority date and not in conflict with the application but cited to understand the principle or theory underlying the invention</li> <li>"X" document of particular relevance; the claimed invention cannot be considered novel or cannot be considered to involve an inventive step</li> <li>"Y" document of particular relevance; the claimed invention cannot be considered to involve an inventive step</li> <li>"Y" document of particular relevance; the claimed invention cannot be considered to involve an inventive step</li> <li>"Y" document of particular relevance; the claimed invention cannot be considered to involve an inventive step</li> <li>"Y" document of particular relevance; the claimed invention cannot be considered novel or cannot be considered novel or cannot be considered to involve an inventive step</li> <li>"Y" document of particular relevance; the claimed invention cannot be considered to involve an inventive step</li> <li>"Y" document of particular relevance; the claimed invention cannot be considered novel or cannot be co</li></ul>							
IV. CERTIFICATION							
		npletion of the International Search	Date of Mailing of this International Search Report				
6 September 1984			Signature of Authorized Officer 20				
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